¹³C NMR QUESTIONS

QUESTION ONE

Work out how many ¹³C peaks you would expect to find in the NMR spectra for the following molecules:

a)
$$\operatorname{CH}_2$$
 CH_2 CH_3 CH_3 CH_3 CH_3

$$\operatorname{CH}_3$$
 CH_2 CH_2 CH_2 CI

d) Br
$$CH_2$$
 CH_2 CH_2 NH_2

e)
$$CH_3$$
 O CH_2 CH_3 CH_2 CH_3 CH_2 CH_3 CH_3 CH_4 CH_3 CH_5 CH_6 CH_7 CH_8 CH_8 CH_8 CH_8 CH_9 CH_9

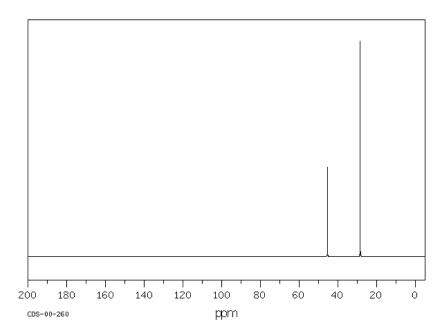
$$CH_3$$
 CH_2-CH_2
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

h)
$$CH_3$$
 C CH_3 CH_2 CH_3 CH_3 CH_3 CH_3

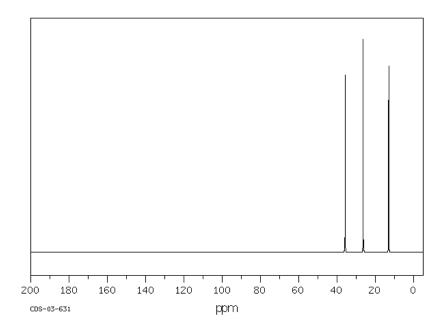
QUESTION TWO

Each of the following ¹³C NMR spectra represents one of the possible isomers of each of the corresponding molecular formulae. For each question draw the chemical structure of the correct isomer.

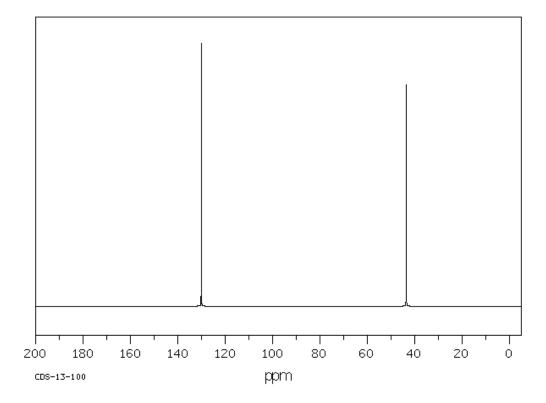
a) C₃H₇Br



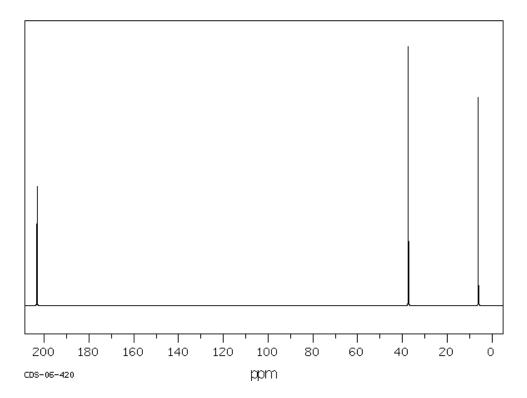
b) C₃H₇Br



c) C₄H₆Cl₂ (contains a C=C double bond)



d) C₃H₆O (contains a C=O bond)



QUESTION THREE

There are four possible structures for <u>an alcohol</u> with the molecular formula $C_4H_{10}O$. Draw the structural formulae for all four alcohols and determine which one belongs to the following ^{13}C NMR spectra

